

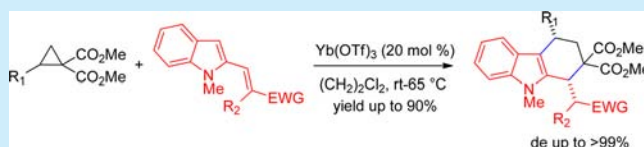
Diastereoselective Synthesis of Functionalized Tetrahydrocarbazoles via a Domino-Ring Opening–Cyclization of Donor–Acceptor Cyclopropanes with Substituted 2-Vinyliindoles

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S Supporting Information

ABSTRACT: A new domino synthetic approach for the synthesis of highly functionalized tetrahydrocarbazoles via DROC of various functionalized DA-cyclopropanes with 2-indolynitroethylene and indole-substituted alkylidene malonate is described. The tetrahydrocarbazoles were obtained with excellent diastereoselectivity having *cis* alignment of the 1,4-appendages across the six-membered carbocyclic ring.



Tetrahydrocarbazole¹ is one of the most important classes of indole-based molecular frameworks. These heterocycles contain a saturated six-membered carbocycle fused to indole at the C-2 and C-3 positions. They are integral parts of several biologically active natural products² and other pharmaceutically relevant synthetic compounds.³

Over the years, several interesting synthetic routes have been devised for the synthesis of substituted tetrahydrocarbazoles and their analogues.^{4,5} Some of the most important strategies include intramolecular allylic arylation,^{4a,b} Friedel–Crafts-type cyclization reactions,^{4c,d} and [4 + 2] cycloaddition of indole derivative bearing an unsaturated functionality at the C-2 or C-3 position with various kinds of dienophiles.⁵ Kerr and co-workers reported an elegant [3 + 3] annulative strategy based upon the reaction of donor–acceptor (DA)-cyclopropanes with 2-alkynylindoles for the synthesis of substituted tetrahydrocarbazoles.⁶ Development of a highly stereoselective route to substituted tetrahydrocarbazoles is still a challenging job and certainly would add value to the overall efforts toward tetrahydrocarbazole synthesis.

DA-cyclopropanes are versatile intermediates in synthetic organic chemistry^{7,8} and have been utilized for the synthesis of a number of organic compounds including several bioactive natural products and drugs. Our recent ongoing research activity and success on DROC of DA-cyclopropanes for the construction of substituted carbocyclic rings⁹ prompted us to investigate the synthesis of diastereomerically pure substituted tetrahydrocarbazoles.

We anticipated that the ring-opening of DA-cyclopropanes with indoles bearing an activated olefin would generate the corresponding substituted malonate anion which would further react with the tethered olefinic moiety via an intramolecular Michael reaction in a domino fashion leading to the diastereoselective formation of tetrahydrocarbazole scaffolds (Scheme 1). Herein, we report a new protocol for the synthesis of highly substituted tetrahydrocarbazoles via DROC of DA-cyclopropanes with substituted 2-vinyliindoles.

Scheme 1. Synthesis of Tetrahydrocarbazoles via DROC of 2-Activated Olefin-Tethered Indoles with DA-cyclopropanes



For this purpose, we have prepared indole substrates with activated olefinic moieties, **2a** and **2b**, following literature reports.¹⁰ At the outset of our study, we performed the reaction of DA-cyclopropane **1a** with 2-indolynitroethylene (**2a**) in the presence of a catalytic amount of Yb(OTf)₃ (20 mol %) as the Lewis acid (LA) catalyst in THF, and the product tetrahydrocarbazole **3a** was obtained in only 14% yield (Table 1, entry 1). In order to optimize the reaction conditions, we screened different solvents and Lewis catalysts. The optimal result came with 1.0 equiv of **1a** and **2a** in 1,2-dichloroethane in the presence of 20% LA, which afforded the tetrahydrocarbazole **3a** in high yield (69%) as a single diastereomer (Table 1, entry 2).

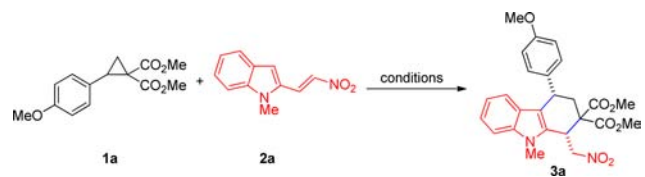
The structure of **3a** was determined by spectroscopic techniques and the relative stereochemistry was confirmed by X-ray crystallographic analysis where aryl and nitro methyl groups were found to possess *cis* appendages (Figure 1).

To generalize this approach, a number of DA-cyclopropanes with a variety of aryl substituents were studied for the DROC with 2-indolynitroethylene **2a**, and the results are shown in Table 2. In the case of the reaction of **2a** with 2-phenylcyclopropanedicarboxylate **1b**, the corresponding tetrahydrocarbazole **3b** was obtained as a single diastereomer in comparable yield. 4-Methylphenylcyclopropanedicarboxylate **1c** behaved similar to **2a** and generated tetrahydrocarbazole **3c**. Fluorinated tetrahydrocarbazole **3d** was also synthesized in high yield when **1d** was reacted with **2a** (Table 2, entry 4).

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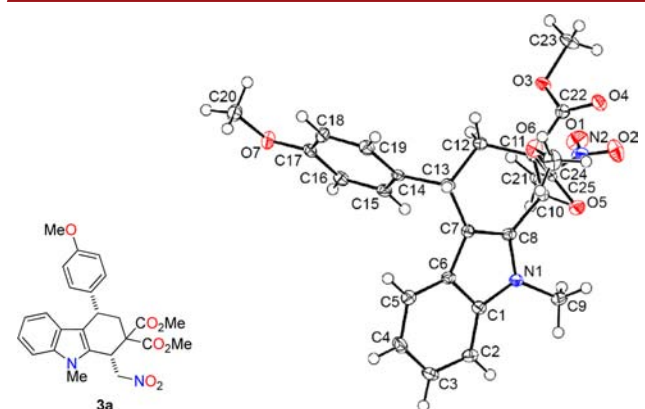
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Table 1. Optimization Study



entry	conditions ^a	time	yield ^b (%)
1	Yb(OTf) ₃ (20 mol %), THF, rt to 60 °C	12 h	14
2	Yb(OTf) ₃ (20 mol %), DCE, rt to 65 °C	30 min	69
3	Yb(OTf) ₃ (1.0 equiv), DCE, rt to 65 °C	20 min	69
4	Yb(OTf) ₃ (20 mol %), DCE, rt	2 days	67
5	Yb(OTf) ₃ (20 mol %), DCM, rt	1 day	58
6	Sc(OTf) ₃ (20 mol %), DCE, rt to 65 °C	45 min	62
7	Zn(OTf) ₂ (20 mol %), DCE, rt to 65 °C	4 h	55
8	Cu(OTf) ₂ (20 mol %), DCE, rt to 65 °C	5 min	10
9	AlCl ₃ (20 mol %), DCE, rt to 65 °C	20 min	25
10	FeCl ₃ (20 mol %), DCE, rt to 65 °C	3 h	30
11	Yb(OTf) ₃ (20 mol %), toluene, rt to 65 °C	50 min	44

^aIn all cases, 1.0 equiv of **1a** was reacted with 1.0 equiv of **2a**. ^bThe product was obtained as a single diastereomer.

Figure 1. X-ray crystal structure of **3a** with 50% thermal ellipsoids.

Similarly, naphthyl and thiophene-yl tetrahydrocarbazoles (**3e** and **3f**, respectively) were generated in high yields via DROC of the corresponding DA-cyclopropanes **1e** and **1f**, respectively, with **2a** (Table 2, entries 5 and 6, respectively).

Next, to extend the scope of our methodology, reactions of several functionalized DA-cyclopropanes **1a–c,f–i** were carried out with indole substituted alkylidene malonate **2b** under the same optimized reaction conditions, and the results are summarized in Table 3.

When the reaction of **1a** was performed with **2b**, the product tetrahydrocarbazole tetraester **3g** was obtained in excellent yield as a single diastereomer (Table 3, entry 1). Similarly, DA-cyclopropanes **1b** and **1c** reacted with **2b** to provide the corresponding products **3h** and **3i**, respectively. Heteroaryl tetrahydrocarbazole tetraesters **3j** and **3k** could also be generated when thiophene-2-yl and 2-furyl cyclopropane dicarboxylates **1f** and **1g** were reacted with **2b**. DA-cyclopropanes having electron-donating substituents on the aromatic rings also afforded the corresponding tetrahydrocarbazoles (**3l,m**) in excellent yields (Table 3, entries 6 and 7, respectively). Halo-substituted DA-cyclopropanes **1j,k** served as excellent substrates for this reaction and produced the corresponding products **3n,o** in good yields. The Michael acceptor **2b** gave the corresponding tetraester products **3g–p** in excellent yields.

Table 2. Reaction of DA-cyclopropanes with 2-Indolylidene malonate **2a**^{a,b}

entry	1	3	time (min)	yield (%)
1			30	69
2			40	69
3			35	65
4			45	66
5			90	64
6			30	65

^aYields of isolated products. ^bThe compounds were obtained as single diastereomers.

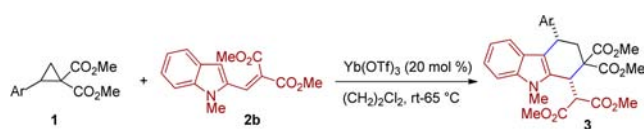
The relative stereochemistry of **3p** was also found to be *cis* as determined by single-crystal X-ray diffraction data (Figure 2).

The scope of our strategy was further extended employing styrylcyclopropanedicarboxylate **1m** as the DA-cyclopropane. When **1m** was reacted with **2a** and **2b**, the corresponding tetrahydrocarbazoles **3q** and **3r**, respectively, were obtained in high yields (Scheme 2, eqs 1 and 2). Although the carbazole **3q** was obtained as a single diastereomer, **3r** was produced as a mixture of diastereomers (dr 4:1).

In the case of the reaction of **1m** with **2a**, a trace amount of the uncyclized product **4** was also isolated. This clearly indicated that the two steps, ring opening and intramolecular Michael addition, take place sequentially in a domino fashion.

Interestingly, when 4-methoxystyrylcyclopropanedicarboxylate **1m** was reacted with **2b**, cyclopentene dicarboxylate **5** was obtained in 90% yield as the sole product; this was generated via intramolecular rearrangement of **1m** upon interaction with the LA catalyst (Scheme 3). It is worth mentioning that construction of substituted cyclopentenones has always been an important goal in organic synthesis.¹¹

The mechanism of the reaction is depicted in Scheme 4. The Lewis acid activates the cyclopropane ring generating a complex **D**, which upon attack of the indole nucleophile generates intermediate **E**. The ring-opened intermediate **E** undergoes attack of its malonate anion moiety to the tethered activated olefin in Michael fashion to produce the tetrahydrocarbazoles. The intermediate **E** can adopt two different cyclohexene like half-chair conformations **F** and **G**. In the conformer **F**, the Michael acceptor adopts a pseudoaxial position and faces a gauche interaction with one of the ester moieties, whereas the Michael acceptor adopts a pseudoequatorial position in the conformer **G** and suffers from a severe gauche interaction with both the ester groups. The more stable conformer **F** is preferred over **G**.

Table 3. Reaction of DA-cyclopropanes with Indole-Substituted Alkylidene Malonate 2b^{a,b}


entry	1	3	time (min)	yield (%)
1			40	88
2			60	85
3			45	86
4			50	75
5			60	87 ^c
6			40	80
7			45	83
8			70	78
9			75	90
10			90	85

^aYields of isolated products. ^bUnless otherwise noted, all the compounds were obtained as single diastereomers. ^cObtained as a 7:2 diastereomeric mixture.

providing tetrahydrocarbazoles with a *cis* orientation of the 1,4-substituents on the cyclohexyl ring.

In conclusion, we have developed an efficient protocol for the synthesis of highly functionalized tetrahydrocarbazoles with

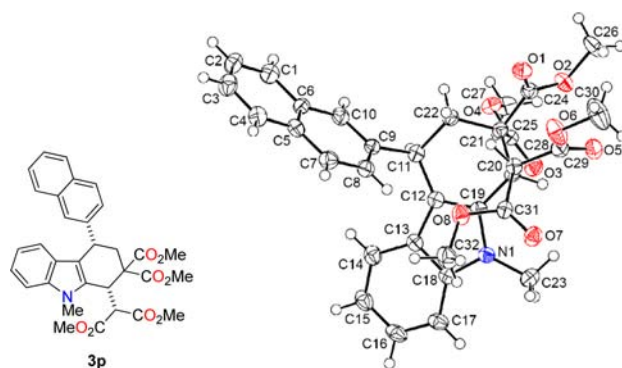
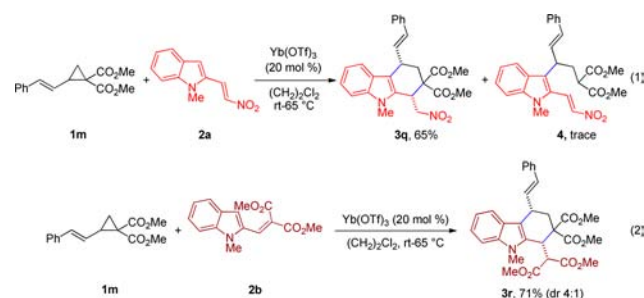
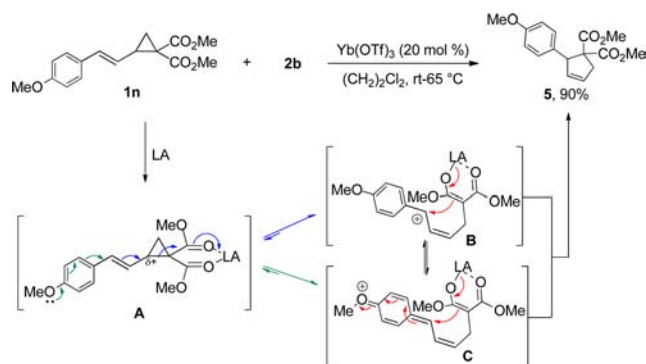


Figure 2. X-ray crystal structure of 3p with 50% thermal ellipsoids.

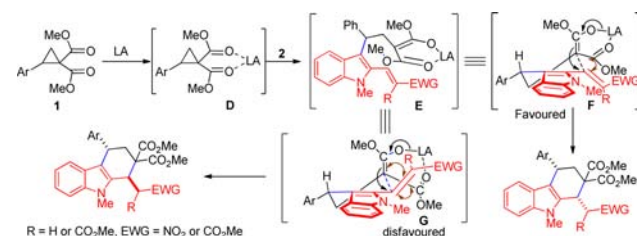
Scheme 2. Synthesis of Styryl-Substituted Tetrahydrocarbazoles 3q and 3r



Scheme 3. Synthesis of Cyclopentene Dicarboxylate via Intramolecular Rearrangement of 4-Methoxystyrylcyclopropanedicarboxylate



Scheme 4. Mechanism of the Reaction and Origin of Diastereoselectivity



excellent diastereoselectivity (de up to 99%) via domino ring-opening cyclization (DROC) of DA-cyclopropanes with 2-indolynitroethylene 2a and indole-substituted alkylidene malonate 2b. The reaction proceeds via a half chairlike transition state where the activated olefin adopts a more stable pseudoaxial

transition state for generation of a six-membered carbocyclic ring with 1,4-*cis* appendages.

■ ASSOCIATED CONTENT

Supporting Information

Detailed experimental procedures, characterization data and X-ray crystallographic analysis of **3a** and **3p**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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■ DEDICATION

Dedicated to Prof. R. N. Mukherjee on the occasion of his 61st birthday.

■ REFERENCES

- (1) (a) Patir, S.; Ertürk, E. *Org. Biomol. Chem.* **2013**, *11*, 2804. (b) Adams, G. L.; Carroll, P. J.; Smith, A. B., III. *J. Am. Chem. Soc.* **2013**, *135*, 519. (c) Wang, W.; Dong, G.; Gu, J.; Zhang, Y.; Wang, S.; Zhu, S.; Liu, Y.; Miao, Z.; Yao, J.; Zhang, W.; Sheng, C. *Med. Chem. Commun.* **2013**, *4*, 353. (d) Bennasar, M.-L.; Zulaica, E.; Ramfrez, A.; Bosch, J. *Tetrahedron* **1999**, *55*, 3117.
- (2) (a) Pandey, G.; Prasanna, C. K. *Org. Lett.* **2011**, *13*, 4672. (b) Jones, S. B.; Simmons, B.; MacMillan, D. W. C. *J. Am. Chem. Soc.* **2009**, *131*, 13606. (c) Jones, S. B.; Simmons, B.; Mastracchio, A.; MacMillan, D. W. C. *Nature* **2011**, *475*, 183. (d) Kozmin, S. A.; Iwama, T.; Huang, Y.; Rawal, V. H. *J. Am. Chem. Soc.* **2002**, *124*, 4628. (e) Ohshima, T.; Xu, Y.; Takita, R.; Shimizu, S.; Zhong, D.; Shibasaki, M. *J. Am. Chem. Soc.* **2002**, *124*, 14546 and references therein. (f) Ueda, H.; Satoh, H.; Matsumoto, K.; Sugimoto, K.; Fukuyama, T.; Tokuyama, H. *Angew. Chem., Int. Ed.* **2009**, *48*, 7600. (g) Nicolaou, K. C.; Dalby, S. M.; Li, S.; Suzuki, T.; Chen, D. Y.-K. *Angew. Chem., Int. Ed.* **2009**, *48*, 7616.
- (3) (a) Nirogi, R. V. S.; Konda, J. B.; Kambhampati, R.; Shinde, A.; Bandyala, T. R.; Gudla, P.; Kandukuri, K. K.; Jayarajan, P.; Kandikere, V.; Dubey, P. K. *Bioorg. Med. Chem. Lett.* **2012**, *22*, 6980. (b) Mittapalli, G. K.; Jackson, A.; Zhao, F.; Lee, H.; Chow, S.; McKelvy, J.; Wong-Staal, F.; Macdonald, J. E. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 6852. (c) Kimura, T.; Hosokawa-Muto, J.; Asami, K.; Murai, T.; Kuwata, K. *Eur. J. Med. Chem.* **2011**, *46*, 5675.
- (4) (a) Bandini, M.; Eichholzer, A. *Angew. Chem., Int. Ed.* **2009**, *48*, 9533. (b) Wu, Q.-F.; Zheng, C.; You, S.-L. *Angew. Chem., Int. Ed.* **2012**, *51*, 1680. (c) Loh, C. C. J.; Raabe, G.; Enders, D. *Chem.—Eur. J.* **2012**, *18*, 13250. (d) Wang, M.-Z.; Zhou, C.-Y.; Che, C.-M. *Chem. Commun.* **2011**, *47*, 1312. (e) Müller, S.; Webber, M. J.; List, B. *J. Am. Chem. Soc.* **2011**, *133*, 18534. (f) Bandini, M.; Gualandi, A.; Monari, M.; Romaniello, A.; Savoia, D.; Tragni, M. *J. Organomet. Chem.* **2011**, *696*, 338. (g) Liu, C.; Widenhoefer, R. A. *Org. Lett.* **2007**, *9*, 1935. (h) Han, X.; Widenhoefer, R. A. *Org. Lett.* **2006**, *8*, 3801. (i) Maertens, F.; Toppet, S.; Hoornaert, G. J.; Compennolle, F. *Tetrahedron* **2005**, *61*, 1715. (j) Jiricek, J.; Blechert, S. *J. Am. Chem. Soc.* **2004**, *126*, 3534.
- (5) (a) Xiao, Y.-C.; Zhou, Q.-Q.; Dong, L.; Liu, T.-Y.; Chen, Y.-C. *Org. Lett.* **2012**, *14*, 5940. (b) Wang, X.-F.; Chen, J.-R.; Cao, Y.-J.; Cheng, H.-G.; Xiao, W.-J. *Org. Lett.* **2010**, *12*, 1140. (c) Liu, Y.; Nappi, M.; Arceo, E.; Vera, S.; Melchiorre, P. *J. Am. Chem. Soc.* **2011**, *133*, 15212. (d) Tan, F.; Li, F.; Zhang, X.-X.; Wang, X.-F.; Cheng, H.-G.; Chen, J.-R.; Xiao,

- W.-J. *Tetrahedron* **2011**, *67*, 446. (e) Gioia, C.; Bernardi, L.; Ricci, A. *Synthesis* **2010**, *1*, 161. (f) Cao, Y.-J.; Cheng, H.-G.; Lu, L.-Q.; Zhang, J.-J.; Cheng, Y.; Chen, J.-R.; Xiao, W.-J. *Adv. Synth. Catal.* **2011**, *353*, 617.
- (6) Grover, H. K.; Lebold, T. P.; Kerr, M. A. *Org. Lett.* **2011**, *13*, 220.
- (7) For reviews, see: (a) Verhé, R.; de Kimpe, N. In *Synthesis and Reactivity of Electrophilic Cyclopropanes*; John Wiley & Sons, Ltd.: New York, 1987. (b) von Angerer, S. In *Carbocyclic Three- and Four-Membered Ring Compounds: Methods of Organic Chemistry* (Houben-Weyl); de Meijere, A., Eds.; Verlag: New York, 1997; Vol. E17c, pp 2121–2153. (c) Reissig, H.-U.; Zimmer, R. *Chem. Rev.* **2003**, *103*, 1151. (d) Mel'nikov, M. Y.; Budynina, E. M.; Ivanova, O. A.; Trushkov, I. V. *Mendeleev Commun.* **2011**, *21*, 293. (e) Tang, P.; Qin, Y. *Synthesis* **2012**, 2969. (f) Cavitt, M. A.; Phun, L. H.; France, S. *Chem. Soc. Rev.* **2014**, *43*, 804. (g) Yu, M.; Pagenkopf, B. L. *Tetrahedron* **2005**, *61*, 321. (h) Lebold, T. P.; Kerr, M. *Pure Appl. Chem.* **2010**, *82*, 1797. (i) Wang, Z. *Synlett* **2012**, 2311. (j) Carson, C. A.; Kerr, M. A. *Chem. Soc. Rev.* **2009**, *38*, 3051. (k) Schneider, T. F.; Kaschel, J.; Werz, D. B. *Angew. Chem., Int. Ed.* **2014**, *53*, 5504.
- (8) For some representative examples of DA-cyclopropane chemistry, see: (a) Pohlhaus, P. D.; Johnson, J. S. *J. Am. Chem. Soc.* **2005**, *127*, 16014. (b) Sibi, M. P.; Ma, Z.; Jasperse, C. P. *J. Am. Chem. Soc.* **2005**, *127*, 5764. (c) Young, I. S.; Kerr, M. A. *J. Am. Chem. Soc.* **2007**, *129*, 1465. (d) Perreault, C.; Goudreau, S.; Zimmer, L.; Charette, A. B. *Org. Lett.* **2008**, *10*, 689. (e) Parsons, A. T.; Smith, A. G.; Neel, A. J.; Johnson, J. S. *J. Am. Chem. Soc.* **2010**, *132*, 9688. (f) Goldberg, A. F. G.; Stoltz, B. M. *Org. Lett.* **2011**, *13*, 4474. (g) Trost, B. M.; Morris, P. J. *Angew. Chem., Int. Ed.* **2011**, *50*, 6167. (h) Rivero, A. R.; Fernández, I.; Sierra, M. A. *Org. Lett.* **2013**, *15*, 4928. (i) Gorbacheva, E. O.; Tabolin, A. A.; Novikov, R. A.; Khomutova, Y. A.; Nelyubina, Y. V.; Tomilov, Y. V.; Ioffe, S. L. *Org. Lett.* **2013**, *15*, 350. (j) Truong, P. M.; Mandler, M. D.; Zavalij, P. Y.; Doyle, M. P. *Org. Lett.* **2013**, *15*, 3278. (k) Fernández-García, J. M.; García-García, P.; Fernández-Rodríguez, M. A.; Pérez-Anesa, A.; Aguilar, E. *Chem. Commun.* **2013**, *49*, 11185. (l) Kaschel, J.; Schmidt, C. D.; Mumby, M.; Kratzert, D.; Stalke, D.; Werz, D. B. *Chem. Commun.* **2013**, *49*, 4403 and references therein. (m) Beyzavi, M. H.; Lentz, D.; Reissig, H.-U.; Wiehe, A. *Eur. J. Org. Chem.* **2013**, 269. (n) Satyanarayana, J.; Rao, M. V. B.; Ila, H.; Junjappa, H. *Tetrahedron Lett.* **1996**, *37*, 3565.
- (9) (a) Ghorai, M. K.; Talukdar, R.; Tiwari, D. P. *Chem. Commun.* **2013**, *49*, 8205. (b) Ghorai, M. K.; Talukdar, R.; Tiwari, D. P. *Org. Lett.* **2014**, *16*, 2204.
- (10) For **2a**: Harley-Mason, J.; Pavel, E. H. *J. Chem. Soc.* **1963**, 2565. For **2b**: Smith, A. B., III; Liu, Z. *Org. Lett.* **2008**, *10*, 4363.
- (11) (a) Maity, S.; Ghosh, S. *Tetrahedron Lett.* **2007**, *48*, 3355. (b) Srikrishna, A.; Khan, I. A.; Babu, R. R.; Sajjanshetty, A. *Tetrahedron* **2007**, *63*, 12616. (c) Nangia, A.; Prasuna, G.; Rao, P. B. *Tetrahedron* **1997**, *53*, 1450. (d) Linghu, X.; Kennedy-Smith, J. J.; Toste, F. D. *Angew. Chem., Int. Ed.* **2007**, *46*, 7671.